

Certificate of Analysis NIM-G GRANITE

SARM 1

CERTIFIED REFERENCE MATERIAL

Distributed by SA BUREAU OF STANDARDS P/Bag X191, Pretoria 0001 Republic of South Africa Prepared by
Council for Mineral Technology (MINTEK)
P/Bag X3015, Randburg 2125
Republic of South Africa

STATUS OF CERTIFICATE

There is no intention of an early revision of this certificate. Further data from improved techniques may necessitate a revision in later years.

2. DATE OF ORIGINAL CERTIFICATION

September 1974.

DATE OF REVISION

May 1979 and June 1984.

4. AVAILABILITY OF OTHER SIZES OF THE MATERIAL

Only 100 g amounts of the powdered material are available.

SOURCE OF THE MATERIAL

The material is from the Bushveld Granite, the acid phase of the Bushveld Complex, in the Transvaal, South Africa. See 18.1.

6. DESCRIPTION OF THE MATERIAL

The material is granitic rock which consists mainly of quartz and K-feldspar with smaller amounts of mica and Na-feldspar; the feldspar, mostly microcline, is highly altered.

INTENDED USE

As a reference material (RM) for the calibration of instruments. As a control sample in the analysis of rock samples of similar type. As a reference material for the development of analytical techniques.

8. STABILITY, TRANSPORT AND STORAGE INSTRUCTIONS

There are no special storage instructions, but care should be taken to avoid undue vibration as this could cause segregation within the bottle.

INSTRUCTIONS FOR THE CORRECT USE OF THE REFERENCE MATERIAL

The material should be dried at 105 $^{\circ}$ C to a constant mass, usually for 2 to 3 hours, before subsamples are taken.

METHOD OF PREPARATION OF THE REFERENCE MATERIAL

Approximately 455 kg of rock were collected with sizes from 0,354 x $10^4 \, \mathrm{cm^3}$ to 7,08 x $10^4 \, \mathrm{cm^3}$.

With the use of various crushing techniques the material was reduced in size until 98 % passed through a sieve with a nominal aperture size of 75 μ m . For full details of the comminution procedure see 18.1.

11. STATE OF HOMOGENEITY

12. CERTIFIED PROPERTY VALUES AND CONFIDENCE LIMITS (where available see 18.5)

Constituent Certified value	Constituent Certified value µg/g	95 % Confidence limits Low High
SiO ₂ 75,70	Ce195	190 203
Al ₂ O ₃	Cr 12 Cu 12	
CaO 0,78	Eu 0,35	0,34 0,47
Na ₂ O 3,36	Ga 27	
K ₂ O 4,99	La 109	100 116
F 0,42	Mn 160	
H ₂ O+ 0,49	Nb 53	
Total Fe as Fe ₂ O ₃ 2,00	Nd 72	67 84
	Pb 40	
	Rb 325	
	Sm 15,8	14,0 18,0
	Sr 10	
	Tb 3,0	2,4 3,5
	Th 51	48 55
	Ti 540	
	Y 143	120 159
	Yb 14,2	12,8 16,0
	Zn 50	*

Zr 300

Uncertified

value

13. UNCERTIFIED/APPROXIMATE PROPERTY VALUES

Uncertified

value

Constituent

	%	μq/q
Fe ₂ O ₃	0.6	Ba 120
MgO		Dy 17
CO ₂		Gd 14
		La 107
		Li 12
		Lu 2
		Ni 8
		Tm 2
		U 15
		V 2
An order of magnitud	de can he	e obtained for the following eler

Constituent

An order of magnitude can be obtained for the following elements from the analytical results given in NIM Report No. 1945-1978 (see 18.3): Ag, As, Au, B, Be, Bi, Cd, Cl, Co, Cs, Hf, Mo, P, S, Sb, Sc, Ta. Similarly an order of magnitude can be obtained for Er, Ho and Pr from MINTEK Report No. M134(1984). See 18.5.

14. VALUES OBTAINED BY INDIVIDUAL LABORATORIES / METHODS

More than 80 laboratories throughout the world have contributed towards the analytical data. See 17.

15. MEASUREMENT TECHNIQUES USED FOR THE CERTIFICATION

Among the techniques used by the contributing laboratories were the following:
Atomic absorption spectrophotometry
Emission spectroscopy
Neutron activation analysis
Spectrophotometry
X-ray fluorescence

Sample dissolution was mainly by acid digestion or fusion. For full details of techniques, etc., see 18.3, 18.4 and 18.5.

TREATMENT OF THE NUMERICAL VALUES

Statistical tests were used to determine outlying results which were then removed from the main population of results. Estimators of central tendency were then used to determine the preferred value. For full details of the statistical treatment of the analytical data see 18.3, 18.4 and 18.5.

NOTE: The certified value is an estimate of the "true" value based upon the best available

See attached list 18. REFERENCES: PREPARATION AND CERTIFICATION PROCEDURES USED IN THIS SAMPLE 18.1 Russell, B G, et al.: Collection and preparation of standard rock samples — Report No. 332-1968 of the National Institute for Metallurgy. 18.2 Russell, B G, et al.: Preliminary report on the analysis of the six NIMROC geochemical standard samples — Report No. 1351-1972 of the National Institute for Metallurgy. 18.3 Steele, T W, et al.: Analysis of the NIMROC reference samples for minor and trace elements — Report No. 1945-1978 of the National Institute for Metallurgy. 18.4 Steele, T W, et al.: Analysis of the NIMROC reference samples for major elements — Report No. 2016-1979 of the National Institute for Metallurgy. 18.5 Hansen, R G and Ring, E J: The NIMROC reference materials; revised values for Thorium, Yttrium, Lanthanum and the rare-earth elements. Randburg, Council for Mineral Technologv. Report No. M134-1984. These reports are available free of charge from the Council for Mineral Technology. SIGNATURE / NAME OF CERTIFYING OFFICER/S 19. Mr R C Mallett

ANALYTICAL SCIENCE DIVISION

COUNCIL FOR MINERAL TECHNOLOGY.

NAMES OF ANALYSTS / INVESTIGATORS / CO-OPERATING LABORATORIES

17.

AUSTRALIA	
Australian Mineral Development Laboratories Commonwealth Scientific and Industrial Research Organization Government Chemical Laboratory	Frewville North Ryde Brisbane
Mineral Research Laboratories (C.S.I.R.O.)	North Ryde
University of Melbourne	Parkville South Bentley
AUSTRIA	John Johns,
University of Vienna, Analysis of Nuclear Raw Materials Division	Vienna
BELGIUM	
Rijksuniversiteit-Gent, Instituut voor Nucleaire Wetenschappen	Ghent
Université de Liège, Institut de Géologie	Liège
CANADA	
Department of Natural Resources	Quebec
Falconbridge Nickel Mines Ltd	Thornhill Ottawa
McGill University, Department of Geological Sciences	Montreal
McMaster University, Department of Geology	Hamilton
Inco Metals Co	Copper Cliff, Ontario
CZECHOSLOVAKIA	
Geological Survey of Czechoslovakia	Prague
FRANCE	
Bureau National de Métrologie	Paris
Bureau de Recherches Géologiques et Minières	Orleans
Centre National d'Etudes et Recherches Céramiques	Paris
Commissariat a l'Energie Atomique	Nancy Chatillon-sous-Bagneux
Rhône-Progil	Aubervilliers
Société Nationale des Pétroles d'Aquitaine	Pau
Université Pierre et Marie Curie, Département de Géologie Appliquée	Paris
GREAT BRITAIN	
British Ceramic Research Association	Stoke-on-Trent
Institute of Geological Sciences	London London
Macaulay Institute for Soil Research	Aberdeen
Open University, Department of Earth Sciences	Milton Keynes
Imperial College of Science and Technology, Department of Geology	London
University of Bristol, Department of Geology	Bristol
University College, Department of Geology	London Staffordshire
University of Leeds, Department of Earth Sciences	Leeds
University of Newcastle-upon-Tyne, Department of Geology	Newcastle-upon-Tyne
University of Reading, Department of Geology	Reading
ISRAEL Coological Survey of Israel	
Geological Survey of Israel	Jerusalem
ITALY	
Instituto di Petrografia, Universitá di Genova	Genova Bologna
JAPAN	
Geological Survey of Japan Kyushu University, Department of Geology	Kawasaki Fukuoka
MEXICO -	
Ciudad Universitaria, Instituto de Geofisica, Torre de Ciencas	Mexico

Lower Hutt

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